

2003


3 (Sem-5) CHM M 3

2016

CHEMISTRY
(Major)

Paper : 5.3

Full Marks : 60

Time : 3 hours

*The figures in the margin indicate full marks
for the questions*

1. Answer the following questions (any seven) : 1×7=7

- (a) Why does colourless aniline on storage turn brown?
- (b) What is 'reductive amination'?
- (c) What happens when acetylene and H_2S are passed over alumina at $400\text{ }^\circ C$?
- (d) Why does diethyl malonate undergo alkylation?
- (e) Why is naphthalene less aromatic than benzene?

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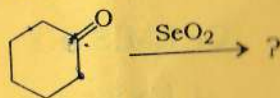
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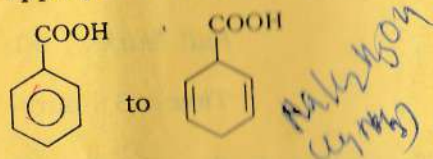
(2)

- (f) Give the product (with name) of the following reaction :



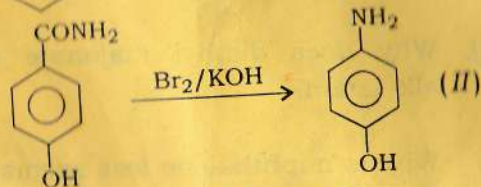
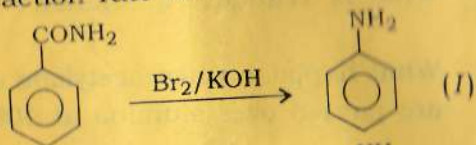
- (g) Define 'conrotatory ring closure'.

- (h) Write the appropriate reagent to convert



2. Answer the following questions (any four) : 2×4=8

- (a) What is enolate anion and how can it be prepared?
- (b) What happens to α -diazoketone when it is heated thermally in presence of Ag₂O? Give the reaction.
- (c) Reaction rate of



(I) is more than (II). Explain.

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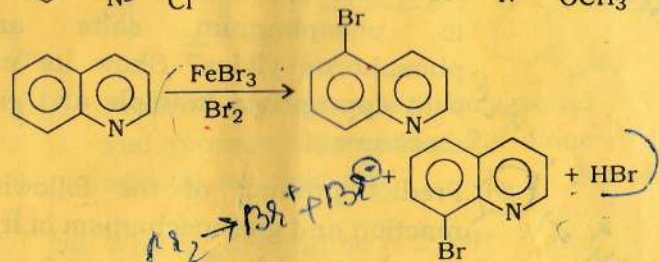
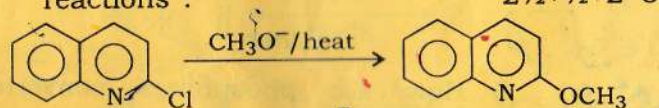
(3)

- (d) Pyridine is basic. Explain.
- (e) Explain the acidic behaviour of nitroalkane.

3. Answer the following questions [any one from (a) and (b) and two from (c), (d) and (e)] :

5+(5×2)=15

- (a) Write Skraup synthesis of quinoline. What is the role of FeSO₄ in this reaction? Explain the reactivity of quinoline with the help of following reactions :



- (b) Define pericyclic reaction. Explain, why electrocyclic reactions are stereospecific, with appropriate examples. 1+4=5
- (c) Convert benzophenone oxime to benzanilide in the presence of PCl₅ and give equation with mechanism. What is the driving force for the 1,2-shift in this reaction? 1+3+1=5

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(4)

(d) Write Nakabayashi mechanism of Clemmensen reduction. Explain why in this reaction amalgamated zinc is used instead of pure zinc. $4+1=5$

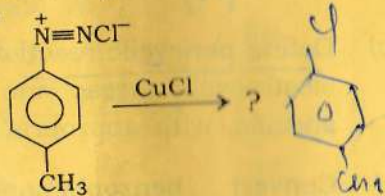
(e) Why do aliphatic nitrocompounds dissolve in aqueous alkali? How can CH_3CN and CH_3NC be prepared? What do you get when they are subjected to acid hydrolysis? $1+2+2=5$

4. Answer the following questions

Either

(a) (i) How are phosphines converted to phosphonium salts and phosphorus ylides? Show its use with appropriate example and give mechanism. $3+2=5$

(ii) Predict product of the following reaction and give mechanism of it: 3

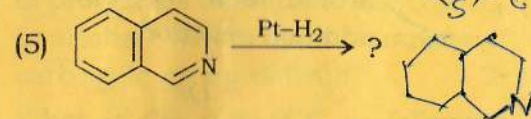
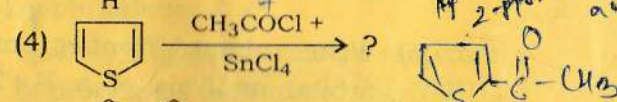
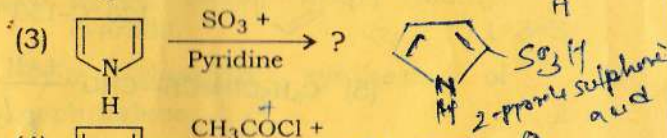
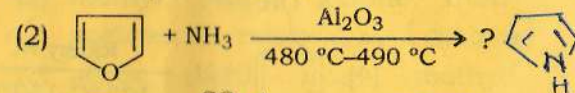
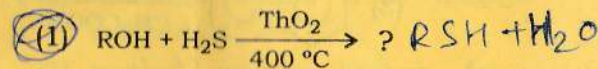


(iii) Why aniline cannot undergo (1) Friedel-Crafts reaction and (2) nitration reaction with HNO_3 ? $1+1=2$

(5)

Or

(b) (i) Write products and the names of the products (wherever possible) for the reactions given below: $1 \times 5 = 5$

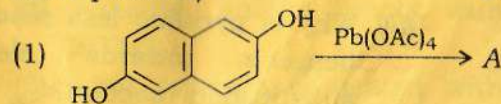


(ii) From ethyl acetoacetate, how would you prepare (1) pentane-2,4-dione and (2) succinic acid? $2+2=4$

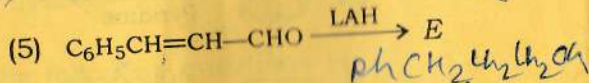
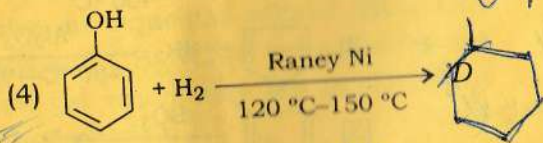
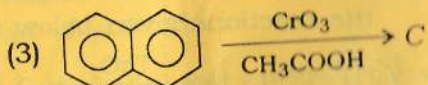
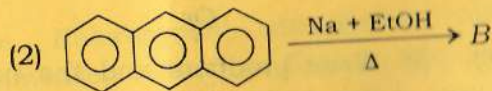
(iii) Why K_b of pyridine is 2.3×10^{-9} and that of pyrrole is 2.5×10^{-4} ? 1

Either

(c) (i) Identify A, B, C, D and E in the following reactions (write name of each product): $1 \times 5 = 5$



(6)



(ii) Write about frontier molecular orbital analysis of a [4+2] cyclo-addition reaction.

Or

(d) (i) Explain 'ion pair mechanism in a solvent cage' with appropriate example of a rearrangement reaction.

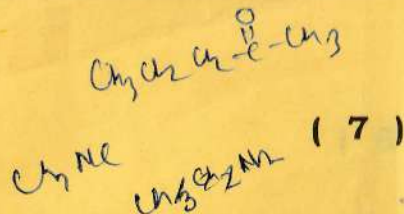
(ii) How can you obtain butanoic acid from diethyl malonate? Write the reaction.

(iii) How can benzene be converted to anthracene with phthalic anhydride? Give reaction.

(iv) Why is pyridine less reactive than benzene towards electrophilic substitution reaction?

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(e) (i) How can you obtain (give reaction)—

(1) dimethyl amine from methyl isocyanide; (2) LAUC

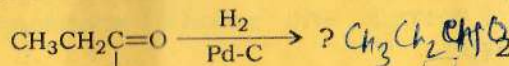
(2) methyl propyl ketone from acetoacetic ester; (3) pentanone

(3) acetic acid from methyl cyanide? 1×3=3

(ii) Write Haworth synthesis of naphthalene.

(iii) How does hydride transfer reduction of propanone take place with NaBH₄ in alcohol? Why is hydride transfer of carboxylic group difficult? 2+1=3

(iv) What is the role of C in following conversion reaction? Give product of the reaction



Or

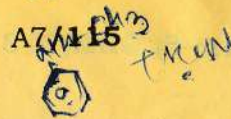
(i) (i) What happens when (give reactions)—

(1) *m*-dinitrobenzene reacts with (NH₄)₂S; N=O

(2) *N*-methyl aniline reacts with HONO; - N-CH₃

(3) ethyl bromide is heated with potassium sulphide? 1×3=3

(C₂H₅)₂SC (Turn Over)

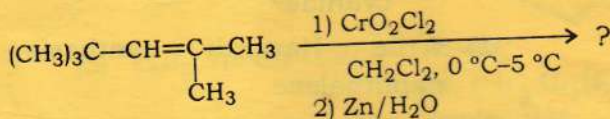


(8)

(ii) Give an example to show the application of benzilic acid rearrangement. 2

(iii) How is primary amine diazotized? Show the mechanism with aniline. 1+2=3

(iv) Complete the following reaction : 1



(v) How is the strong acidity of the reagent, pcc, generally controlled? 1

Handwritten scribbles and symbols, possibly including a plus sign and some illegible characters.